## Novel Heterocyclic Cage Compounds from 2-Methylthiofurans

## Hsien-Jen Wu,\* Fang-Jung Huang and Chu-Chung Lin

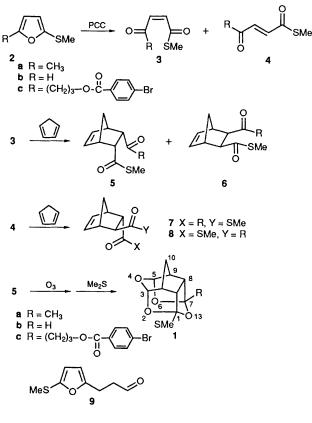
Department of Applied Chemistry, National Chiao-Tung University, Hsinchu, Taiwan, Republic of China

Some novel heterocyclic cage compounds **1a–1c** were synthesized from the corresponding 2-methylthiofurans **2a–2c** in a short sequence.

There is considerable interest in the synthesis of cage compounds,<sup>1</sup> including heterocyclic cage compounds.<sup>2</sup> We report here the synthesis of some novel heterocyclic cage compounds **1a–1c**, which possess four oxygen atoms in the framework, in three steps from the corresponding 2-methyl-thiofurans **2a–2c**.

Metallation<sup>3</sup> of 2-methylfuran with n-butyllithium followed by addition of dimethyldisulphide gave 2-methylthio-5methylfuran **2a** in 85% yield. Oxidation of **2a** with two equivalents of pyridinium chlorochromate (PCC) in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 2 h gave a single product **3a** in 70% yield. A longer oxidation reaction time (24 h) gave the cis-isomer 3a and the *trans*-isomer 4a in a ration of 1:2. Reaction of the cis-isomer 3a with cyclopentadiene at room temperature gave the *endo* adduct 5a as the major product and the *exo* adduct 6a as the minor product in a ratio of 6:1 in 80%yield. Reactions of the *trans*-isomer 4a with cyclopentadiene at room temperature gave the adducts 7a and 8a in a ratio of 1:1 in 80% yield. Compounds 5b, 6b, 7b and 8b were synthesized from 2-methylthiofuran 2b in a similar sequence, Scheme 1.

Reaction of **2b** with acrolein in glacial acetic acid at 60 °C gave the Michael adduct **9**, which following reduction with NaBH<sub>4</sub> and esterification with *p*-bromobenzoyl chloride gave



Scheme 1

compound 2c in 55% overall yield. Compound 5c was synthesized from 2c via a similar sequence as 5a from 2a and 5b from 2b, Scheme 1.

Ozonolysis of compounds 5a, 5b and 5c, all of which have cis-endo stereochemistry, in CH2Cl2 at -78 °C followed by reduction with dimethylsulphide gave the corresponding novel heterocyclic cage compounds 1a, 1b and 1c in 60-68% yields, Scheme 1. The IR spectra lacked the carbonyl absorptions. The <sup>1</sup>H NMR spectrum<sup>†</sup> of **1a** showed two doublets at  $\delta$  5.58 and 5.52 for the two acetal protons on C-3 and C-5, and a singlet at  $\delta$  2.21 for the methylthic protons. The absorption at  $\delta$  2.09 singlet for the methyl ketone protons of 5a shifted to  $\delta$ 1.57 for the angular methyl protons of 1a. The <sup>13</sup>C NMR spectrum lacked any carbonyl absorption and displayed two singlets at  $\delta$  121.9 and 117.7 for the quaternary carbons C-1 and C-7 of compound 1a. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of 1b

and 1c revealed that both compounds 1b and 1c possess the same skeleton as 1a.<sup>†</sup>

In order to understand the effect of the stereochemistry of the Diels-Alder adducts 5-8 on the formation of cage compounds 1, ozonolysis reactions of compounds 6a, 7a and 8a were also performed. No detectable amount of cage compound 1a was formed in either ozonolysis reactions of 6a or the mixture of 7a and 8a. Thus, only the isomers with *cis-endo* stereochemistry could give the corresponding heterocyclic cage compounds.

We thank the National Science Council of the Republic of China for financial support.

Received, 15th February 1991; Com. 1/007311

## References

- 1 P. E. Eaton and T. W. Cole, Jr., J. Am. Chem. Soc., 1964, 86, 962, 3157; P. E. Eaton, R. A. Hudson and C. Giordano, J. Chem. Soc., Chem. Commun., 1974, 978; P. E. Eaton, L. Cassar, R. A. Hudson and D. R. Hwang, J. Org. Chem., 1976, 41, 1445; P. E. Eaton, Y. S. Or and S. J. Branca, J. Am. Chem. Soc., 1981, 103, 2134; A. P. Marchand and D. S. Reddy, J. Org. Chem., 1984, 49, 4078; A. P. Marchand and D. S. Reddy, J. Org. Chem., 1985, 50, 724; A. P. Marchand and A. H. Wu, J. Org. Chem., 1986, 51, 1897; L. A. Paquette, R. J. Ternansky and D. W. Balogh, J. Am. Chem. Soc., 1982, 104, 4502; G. Mehta, K. S. Rao, K. Venkatesan and M. M. Bhadbhade, J. Chem. Soc., Chem. Commun., 1981, 755; L. A. Paquette, Top. Curr. Chem., 1979, **79**, 41.
- 2 K. W. Shen, J. Am. Chem. Soc., 1971, 93, 3064; E. L. Allred and B. R. Beck, Tetrahedron Lett., 1974, 437; G. Mehta and M. S. Nair, J. Chem. Soc., Chem. Commun., 1983, 439; A. P. Marchand and A. H. Wu, J. Org. Chem., 1986, 51, 1897.
  3 H. W. Gschwend and H. R. Rodriguez, Org. React., 1979, 26, 31.

+ Spectral data for cage compounds 1a: highly viscous liquid, IR,  $v_{max}$ (neat) 1050 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 100 MHz), δ 5.58 (1H, d, J 6.6 (near) 1050 cm<sup>-1</sup>, <sup>1</sup>H NMR (CDC<sub>13</sub>, 100 MHz), 6 3.38 (1H, d, *J* 6.6 Hz), 5.52 (1H, d, *J* 6.6 Hz), 3.59 (1H, dd,  $J_1$  7.9,  $J_2$  7.6 Hz), 3.23 (1H, dd,  $J_1$  7.9 Hz,  $J_2$  7.8 Hz), 2.95 (2H, m), 2.21 (3H, s), 1.95–1.85 (2H, m), 1.57 (3H, s); <sup>13</sup>C NMR CDCl<sub>3</sub>, 25.0 MHz),  $\delta$  121.9(s), 117.7(s), 103.4(d), 102.6(d), 59.5(d), 56.5(d), 45.4(d), 55.1(d), 20.6(d), 20.4 (d), 20.4 (d), 20.6(d), 50.5(d), 56.5(d), 56. 45.1(d), 28.6(t), 24.2(q), 12.6(q); high resolution mass  $(C_{11}H_{14}O_4S)$ 242.0609 (calcd. 242.0613). 1b: IR,  $v_{max}$  (neat) 1050 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 100 MHz),  $\delta$  5.88 (1H, d, J 4.9 Hz), 5.58 (1H, d, J 6.6 Hz),  $\delta$  5.60 (1H, d, J 4.9 Hz),  $\delta$  5.80 (2H, d, J 4.9 Hz), \delta 5.80 (2H, d, J 4.9 Hz), 5.52 (1H, d, J 6.6 Hz), 3.52 (2H, m), 2.92 (2H, m), 2.27 (3H, s), 1.92 (2H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 25.0 MHz), 8 122.6(s), 110.0(d), 104.0(d), 102.7(d), 58.6(d), 53.4(d) 45.3(d), 44.9(d), 28.9(t), 12.6(q); high resolution mass ( $C_{10}H_{12}O_4S$ ) 228.0463 (calcd. 228.0456). **1c**: IR,  $v_{max}$  (KBr)1720, 1595, 1280, 1050 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 100 MHz), δ 7.89 (2H, d, J 8.5 Hz), 7.57 (2H, d, J 8.5 Hz), 5.60 (1H, d, J 6.5 Hz), 5.54 (1H, d, J 6.5 Hz), 4.36 (2H, t, J 3.2 Hz), 3.55 (1H, dd, J<sub>1</sub> 8.3 Hz,  $J_2$  7.5 Hz), 3.24 (1H, dd,  $J_1$  8.3H,  $J_2$  7.6 Hz), 2.95 (2H, m), 2.23 (3H, s), 2.05–1.80 (6H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 25.0 MHz),  $\delta$  165.1(s), 131.1(d)(2C), 130.6(d)(2C), 128.6(s), 127.4(s), 122.1(s), 119.4(s), 103.6(d), 102.7(d), 64.5(t), 59.4(d), 55.4(d), 45.6(d), 45.3(d), 33.9(t), 28.8(t), 23.4(t), 12.7(q); high resolution mass (C<sub>20</sub>H<sub>21</sub>O<sub>6</sub>SBr) 470.0245, 468.0240 (calcd. 470.0218, 468.0240).